Concise Assembly of the Polycyclic Frameworks Associated with the Hapalindole and Fischerindole Alkaloids

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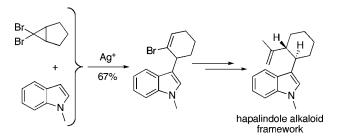
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ABSTRACT

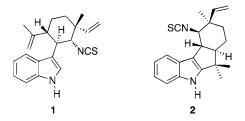


Reaction of *N*-methylindole (4) with 6,6-dibromobicyclo[3.1.0]hexane (5) in the presence of silver tetrafluoroborate affords conjugate 7 in 67% yield. This product can be readily elaborated to compounds 12b and 13b which embody the polycyclic frameworks associated with members of the hapalindole and fischerindole classes of alkaloids. The chiral-auxiliary-substituted 6,6-dibromobicyclo[3.1.0]hexanes 21 and 22 react with indole to give adducts likely to be useful in the enantioselective total synthesis of the title alkaloids.

As part of a continuing program¹ to exploit *gem*-dihalocyclopropanes in the synthesis of natural products, we sought to identify carbon-centered (or C) nucleophiles capable of trapping the allylic cations derived from electrocyclic ringopening of these readily available three-membered-ring compounds. While some seminal work along such lines has been reported by Gassman and co-workers,² the range of C-nucleophiles known to participate is limited to simple

(2) Gassman, P. G.; Tan, L.; Hoye, T. R. Tetrahedron Lett. **1996**, *37*, 439.

10.1021/ol062020x CCC: \$33.50 © 2006 American Chemical Society Published on Web 09/23/2006 alkenes and aromatics.³ Herein, therefore, we report that various indoles also engage, as nucleophiles, in these trapping processes. In addition, we also detail the exploitation of adducts derived from such reactions in the rapid assembly of the polycyclic frameworks associated with the biologically and structurally interesting hapalindole and fischerindole classes of alkaloids,⁴ representative examples being (+)-hapalindole Q (1)^{4b} and (-)-12-*epi*-fischerindole U isothiocyanate (2).^{4e} These types of natural products, which were

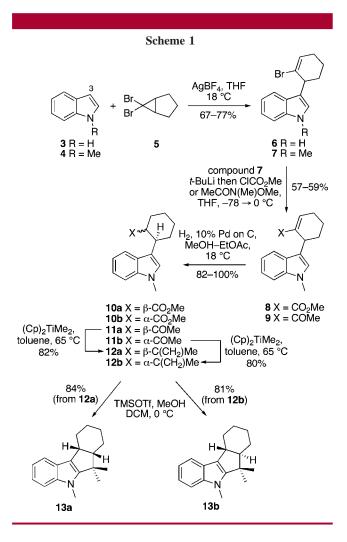


first isolated by Moore and co-workers and can be found in a variety of organisms, most notably blue-green algae,⁴ display anti-bacterial, anti-algal and anti-mycotic activities.⁴

^{(1) (}a) Banwell, M. G.; Gable, R. W.; Peters, S. C.; Phyland, J. R. J. Chem. Soc., Chem. Commun. 1995, 1395. (b) Banwell, M.; Edwards, A.; Harvey, J.; Hockless, D.; Willis, A. J. Chem. Soc., Perkin Trans. 1 2000, 2175. (c) Banwell, M. G.; Harvey, J. E.; Hockless, D. C. R.; Wu, A. W. J. Org. Chem. 2000, 65, 4241. (d) Banwell, M. G.; Ebenbeck, W.; Edwards, A. J. J. Chem. Soc., Perkin Trans. 1 2001, 114. (e) Banwell, M. G.; Harvey, J. E.; Jolliffe, K. A. J. Chem. Soc., Perkin Trans. 1 2001, 2002. (f) Banwell, M. G.; Edwards, A. J.; Jolliffe, K. A.; Smith, J. A.; Hamel, E.; Verdier-Pinard, P. Org. Biomol. Chem. 2003, 1, 296. (g) Taylor, R. M. Aust. J. Chem. 2003, 56, 631. (h) Banwell, M. G.; Sydnes, M. O. Aust. J. Chem. 2004, 57, 537. (i) Stanislawski, P. C.; Willis, A. C.; Banwell, M. G. Org. Lett. 2006, 8, 2143. For a review of some of our group's research efforts in this area see: (j) Banwell, M. G.; Beck, D. A. S.; Stanislawski, P. C.; Sydnes, M. O.; Taylor, R. M. Curr. Org. Chem. 2005, 9, 1589.

Such properties may be attributable to the compounds' capacities to inhibit RNA polymerase.⁵ Accordingly, these alkaloids have stimulated various synthetic studies over the last 15 years or so and total syntheses have been reported by the groups of Natsume, Albizati, Fukuyama, Kerr, and Baran.⁶

Our approach to the title frameworks is shown in Scheme 1 and involved, in the opening stages, an investigation of



the reaction of the parent indole (3) and its *N*-methylated derivative (4) with *gem*-dibromocyclopropane 5, the last compound being readily available through dibromocarbene addition to cyclopentene. In the event, each of indoles 3 and 4 reacted smoothly at 18 °C with equimolar quantities of

(5) (a) Doan, N. T.; Rickards, R. W.; Rothschild, J. M.; Smith, G. D. J. Appl. Phycol. **2000**, *12*, 409. (b) Doan, N. T.; Stewart, P. R.; Smith, G. D. FEMS Microbiol. Lett. **2001**, *196*, 135.

cyclopropane 5 in the presence of silver tetrafluoroborate to give adducts 6 (77%) and 7 (67%), respectively. These products clearly arise through nucleophilic attack of C-3 of the indoles on the allylic cation derived from electrocyclic ring-opening of the compound 5. Reaction of cyclohexenyl bromide 7 with 2 molar equiv of tert-butyllithium and trapping of the ensuing cyclohexenyllithium with either methyl chloroformate or the Weinreb amide of acetic acid7 then gave the corresponding α,β -unsaturated ester 8⁸ (57%) and methyl ketone 9 (59%), respectively.⁹ Subjection of the former product to hydrogenation, at 1 atm, using 10% Pd on C as catalyst gave a ca. 1:1 and chromatographically separable mixture of the cis- and trans-1,2-disubstituted cyclohexanes 10a (51%) and 10b (49%), respectively. Analogous treatment of the acylated alkene 9 afforded a chromatographically separable mixture of compounds 11a (19%) and **11b** (63%). Methylenation of compound **11a** with the Petasis reagent¹⁰ afforded the alkene 12a (82%) that upon treatment with TMSOTf underwent cyclization to give the tetracyclic compound $13a^8$ (84%) embodying the *cis*-ring fused variant on the fischerindole framework. Conversion of isomeric ketone 11b into the corresponding and targeted alkene 12b, which incorporates key elements of natural products such as hapalidoles C, D, E, F, and Q,4,6 was accomplished in an efficient manner (80%) by using the Petasis reagent. Furthermore, subjection of the latter compound to what might be regarded as a biomimetic cyclization reaction^{6d,g} using TMSOTf afforded the tetracyclic indole 13b⁸ (81%) now embodying the framework associated with the fischerindoles.¹¹

The strategy detailed above can also be exploited in the construction of enantiopure indole-substituted cyclohexenes incorporating additional functionality likely to be useful in developing total syntheses of the title alkaloids. The two *gem*-dibromocyclopropanes used in examining this particular aspect of the present studies were prepared by the pathway shown in the early parts of Scheme 2. Thus, oxidation of commercially available 1,6-heptadien-4-ol (diallyl alcohol) to the previously reported¹² diallyl ketone (**14**) was accomplished (90%) with pyridinium chlorochromate (PCC)¹³

⁽³⁾ Dehmlow, E. V.; Schönefeld, J. Justus Liebigs Ann. Chem. 1971, 744, 42.

^{(4) (}a) Moore, R. E.; Cheuk, C.; Patterson, G. M. L. J. Am. Chem. Soc.
1984, 106, 6456. (b) Moore, R. E.; Cheuk, C.; Yang, X.-Q. G.; Patterson, G. M. L.; Bonjouklian, R.; Smitka, T. A.; Mynderse, J. S.; Foster, R. S.; Jones, N. D.; Swartzendruber, J. K.; Deeter, J. B. J. Org. Chem. 1987, 52, 3773. (d) Park, A.; Moore, R. E.; Patterson, G. M. L. J. Org. Chem. 1987, 52, 3773. (d) Park, A.; Moore, R. E.; Patterson, G. M. L. Tetrahedron Lett. 1992, 33, 3257. (e) Stratmann, K.; Moore, R. E.; Bonjouklian, R.; Deeter, J. B.; Patterson, G. M. L.; Shaffer, S.; Smith, C. D.; Smitka, T. A. J. Am. Chem. Soc. 1994, 116, 9935.

^{(6) (}a) Muratake, H.; Kumagami, H.; Natsume, M. *Tetrahedron* **1990**, *46*, 6351 and references therein. (b) Sakagami, M.; Muratake, H.; Natsume, M. *Chem. Pharm. Bull.* **1994**, *42*, 1393. (c) Vaillancourt, V.; Albizati, K. F. *J. Am. Chem. Soc.* **1993**, *115*, 3499. (d) Fukuyama, T.; Chen, X. *J. Am. Chem. Soc.* **1994**, *116*, 3125. (e) Kinsman, A. C.; Kerr, M. A. *Org. Lett.* **2001**, *3*, 3189. (f) Kinsman, A. C.; Kerr, M. A. *J. Am. Chem. Soc.* **2003**, *125*, 14120. (g) Baran, P. S.; Richter, J. M. *J. Am. Chem. Soc.* **2004**, *126*, 7450.

⁽⁷⁾ Oster, T. A.; Harris, T. M. Tetrahedron Lett. 1983, 24, 1851.

⁽⁸⁾ The structure of this compound has been confirmed by single-crystal X-ray analysis. (CCDC reference numbers: **8**, 610988; **13a**, 610989; **13b**, 610990.)

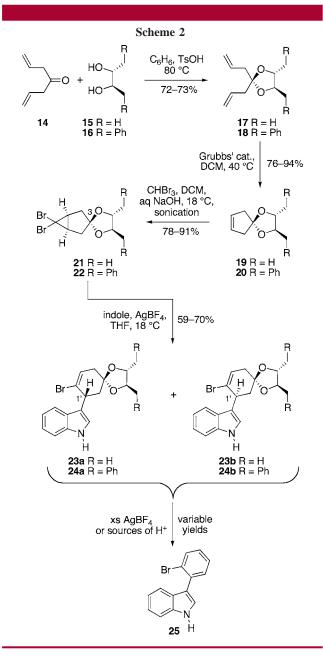
⁽⁹⁾ Attempts to effect analogous conversions of compound **6** through its reaction with 3.0 molar equiv of *t*-BuLi then either acetyl chloride or methyl chloroformate only resulted in the formation of the *N*-acyl or *N*-carbomethoxy derivatives of the starting material. Similar treatment of the readily derived *N*-Boc derivative of compound **6** gave the same products while its *N*-Ts congener afforded low yields (ca. 30%) of the desired cyclohexenes, viz. the *N*-Ts analogues of compounds **8** and **9**.

⁽¹⁰⁾ Petasis, N. A.; Bzowej, E. I. J. Am. Chem. Soc. **1990**, 112, 6392.

⁽¹¹⁾ Methods are available for the demethylation of *N*-methylindoles see, for example: Nakatsuka, S.; Asano, O.; Goto, T. *Heterocycles* **1986**, 24, 2791.

⁽¹²⁾ Dreyfuss, M. P. J. Org. Chem. 1963, 28, 3269.

⁽¹³⁾ Corey, E. J.; Suggs, J. W. Tetrahedron Lett. 1975, 2647.



and the latter compound reacted, under conditions specified by Dolbier and Garza,¹⁴ with commercially available (2R,3R)butane-2,3-diol (**15**) in the presence of catalytic amounts of *p*-TsOH to give the acetal **17** (73%). An analogous process involving diol **16** afforded the diphenylated system **18** (72%). Subjection of each of these compounds to ring closing

metathesis using Grubbs' second-generation catalyst¹⁵ then gave the corresponding cyclopentenes 19 (76%) and 20 (94%). Dibromocarbene addition to the double bonds of the latter compounds, under conditions defined by Xu and Brinker,¹⁶ then afforded the corresponding and targeted ringfused gem-dibromocyclopropanes 21 (91%) and 22 (78%). Reaction of the first of these products, viz. compound 21, with indole in the presence of silver tetrafluoroborate afforded a 1:1 mixture of adducts 23a and 23b (59% combined yield) which could be separated by HPLC techniques. Upon subjecting congener 22, incorporating the second chiral auxiliary, to the same conditions a ca. 1:1 mixture of compounds 24a and 24b (70% combined yield) was obtained. Each of the four adducts 23a, 23b, 24a, and 24b was prone to aromatization upon prolonged exposure to acidic conditions and thereby affording 3-(o-bromophenyl)indole (25). Interestingly, the specific rotations of each of the diastereoisomeric compounds 23a and 23b proved to be of a similar magnitude but opposite sign. The same is true for congeners 24a and 24b. The illustrated stereochemical assignments at C-1' within each of these adducts must be regarded as tentative at this stage.

The studies described above highlight the ability of indoles to act as C-nucleophiles capable of trapping π -allyl cations derived from electrocyclic ring-opening of ring-fused *gem*dibromocyclopropanes. The ensuing adducts are obtained in good yield and can be used for the rapid assembly of analogues of the hapalindole and fischerindole classes of alkaloid. Clearly, the diastereoselectivity of these processes is not strongly influenced by the presence of chiral auxiliaries attached to C3 of the 6,6-dibromobicyclo[3.1.0]hexane framework. Work aimed at exploiting the results reported above in various contexts, including the total synthesis of the title alkaloids, is now underway in our laboratories. Details will be reported in due course.

Supporting Information Available: Preparation and characterization of 5–14 and 16–25, ¹H or ¹³C NMR spectra for 6–13 and 17–24, and X-ray crystallographic data for 8, 13a, and 13b. This material is available free of charge via the Internet at http://pubs.acs.org.

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 ⁽¹⁴⁾ Dolbier, W. R., Jr.; Garza, O. T. J. Org. Chem. 1978, 43, 3848.
 (15) Louie, J.; Bielawski, C. W.; Grubbs, R. H. J. Am. Chem. Soc. 2001,

^{123, 11312.}

⁽¹⁶⁾ Xu, L.; Brinker, U. H. In *Synthetic Organic Sonochemistry*; Luche J.-L., Ed.; Plenum Press: New York, 1998; pp 344–345.